# **BRIEF COMMUNICATION**

# DISPERSION OF A SOLUTE IN A MICROPOLAR PIPE FLOW

GOODARZ AHMADI Pahlavi University, Shiraz, Iran

## 1. INTRODUCTION

Taylor (1953, 1954a, b) discussed the dispersion of soluble matter in the viscous, incompressible laminar flow of a fluid in a circular pipe. The case of the dispersion of a solute in non-Newtonian fluid flow in a circular pipe was discussed by Fan & Hwang (1965). Recently Soundalgekar (1971) studied the effect of couple stresses on the dispersion of a solute in a channel flow.

In the present work the dispersion of a soluble matter in a micropolar pipe flow is considered and the effective Taylor diffusivity is calculated. The effects of various dimensionless parameters on the effective diffusivity are discussed.

### 2. BASIC EQUATIONS

The expression for the velocity in a fully developed pipe flow as derived by Eringen (1966), is given by

$$\frac{u}{u_o} = 1 - \rho^2 + \frac{k}{\mu + k} \frac{1}{\lambda} \frac{I_o(\lambda)}{I_1(\lambda)} \left[ \frac{I_o(\lambda\rho)}{I_o(\lambda)} - 1 \right],$$
[1]

where

$$u_o = \frac{-R}{2(2\mu+k)} \frac{\mathrm{d}p}{\mathrm{d}x}, \quad \lambda = \left(\frac{2\mu+k}{\mu+k}\frac{k}{\gamma}\right)^{1/2} R, \quad \rho = \frac{r}{R}.$$
 [2]

Here  $u_o$  is the maximum velocity, R is the radius of the pipe and dp/dx is pressure gradient along the pipe.  $\mu$ , k and  $\gamma$  are the coefficients of viscosity.  $I_o()$  and  $I_1()$  are the modified Bessell functions of the zeroth and first order, respectively. Also note that the dimension of k is the same as  $\mu$  but the dimension of  $\gamma$  is viscosity times length square. Hence  $\lambda$  is a dimensionless quantity.

The concentration c of the solute diffusing in the fluid satisfies the following equations

$$\frac{\partial c}{\partial t} + u \frac{\partial c}{\partial x} = D \left[ \frac{\partial^2 c}{\partial x^2} + \frac{1}{r} \frac{\partial}{\partial r} \left( r \frac{\partial c}{\partial r} \right) \right],$$
[3]

where D is the molecular diffusion coefficient.

Following Taylor (1953, 1954a, b) we assume that the axial diffusion is much smaller than the radial diffusion.

Using the dimensionless quantities

$$\xi = (x - \bar{u}t)/L, \quad \theta = tL/\bar{u}, \quad [4]$$

the diffusion equation [3] in a frame moving with the average velocity  $\bar{u}$  becomes

$$\frac{\bar{u}}{L}\frac{\partial c}{\partial \theta} + \frac{w}{L}\frac{\partial c}{\partial \xi} = \frac{D}{R^2}\frac{1}{\rho}\frac{\partial}{\partial \rho}\left(\rho\frac{\partial c}{\partial \rho}\right).$$
[5]

where

$$w/u_o = (u - \bar{u})u_o$$

$$=\frac{1}{2}-\rho^{2}+\frac{k}{\mu+k}\frac{1}{\lambda}\frac{1}{I_{1}(\lambda)}\left[I_{o}(\lambda\rho)-\frac{2}{\lambda}I_{1}(\lambda)\right],$$
[6]

and where L is a given length along the flow direction.

Assume Taylor's limiting conditions to be applicable, then the partial equilibrium may be assumed in any cross section of the pipe and c then satisfies:

$$\frac{1}{\rho}\frac{\partial}{\partial\rho}\left(\rho\frac{\partial c}{\partial\rho}\right) = \frac{R^2}{DL}w\frac{\partial c}{\partial\xi}.$$
[7]

Substituting for w from [6], multiplying by  $\rho$ , integrating once, then dividing by  $\rho$  and integrating again we find

$$c = \frac{R^2 u_o}{DL} \left\{ \frac{\rho^2}{8} - \frac{\rho^4}{16} + \frac{k}{\mu + k} \frac{1}{\lambda^2} \frac{1}{I_1(\lambda)} \left[ \frac{1}{\lambda} I_o(\lambda \rho) - \frac{\rho^2}{2} I_1(\lambda) \right] \right\} \frac{\partial c}{\partial \xi} + c_o, \qquad [8]$$

where c satisfies the boundary condition

$$\left. \frac{\partial c}{\partial \rho} \right|_{\rho=1} = o, \tag{9}$$

and  $c_o$  is a constant which can be determined from the entry condition.

Now the volume rate of the transport of the solute across a section of the pipe is given by

$$Q = \int_{o}^{R} 2\pi r \ cw \ dr.$$
 [10]

Inserting for c and w

$$Q = \frac{2\pi R^4 u_o^2}{DL} F\left(\frac{k}{\mu}, \lambda\right) \frac{\partial c}{\partial \xi};$$
[11]

488

λ Κ/μ	1	2	3	4
0	2.20	2.20	2.20	2.20
1	86.49	3.79	1.40	0.90
2	143.40	4.43	1.11	0.59
3	177.26	4.77	0.96	0.39
4	199.30	4.98	0.87	0.27
5	214.72	5.13	0.80	0.19
6	226.08	5.23	0.76	0.13
7	234.80	5.31	0.73	0.09
8	241.69	5.37	0.70	0.06
9	247.28	5.42	0.68	0.04

Table 1. Values of  $F(k/\mu, \lambda) \times 10^{+3}$ .

where

$$F\left(\frac{k}{\mu},\lambda\right) = \int_{o}^{1} \rho \left\{\frac{1}{2} - \rho^{2} + \frac{k}{\mu+k}\frac{1}{\lambda}\frac{1}{I_{1}(\lambda)}\left[I_{o}(\lambda\rho) - \frac{2}{\lambda}I_{1}(\lambda)\right]\right\}$$
$$\left\{\frac{\rho^{2}}{8} - \frac{\rho^{4}}{16} + \frac{k}{\mu+k}\frac{1}{\lambda^{2}}\frac{1}{I_{1}(\lambda)}\left[\frac{1}{\lambda}I_{o}(\lambda\rho) - \frac{\rho^{2}}{2}I_{1}(\lambda)\right]\right\}d\rho. \quad [12]$$

On comparing [11] with Fick's law of diffusion, one can show that the solute is dispersed relative to a plane moving with the mean speed of the flow with an effective Taylor diffusion coefficient  $D^*$  given by

$$D^* = \frac{2R^2 u_o^2}{D} F\left(\frac{k}{\mu}, \lambda\right).$$
 [13]

The numerical values of  $F(k/\mu, \lambda)$  are given in table 1.

#### 3. CONCLUSION

The effective Taylor diffusion coefficient is found to be governed by two nondimensional parameters  $k/\mu$  and  $\lambda$ . For  $\lambda = 1$ ,  $F(k/\mu, \lambda)$  increases rapidly with  $k/\mu$ . The rate of increase is slower for  $\lambda = 2$ . However, for  $\lambda = 3$  the effective diffusivity decreases with an increase in  $k/\mu$ . For large values of  $\lambda$  and  $k/\mu$  (>30),  $F(k/\mu, \lambda)$  remains approximately constant (=2.2 × 10<sup>-3</sup>). In this range the effective diffusivity becomes independent of  $\lambda$  and  $k/\mu$  and approaches that of a simple viscous fluid.

## REFERENCES

ERINGEN, A. C. 1966 Theory of micropolar fluids. J. Math. Mech. 14, 1-18.

- FAN, L. T. & HWANG, W. S. 1965 Dispersion of Ostwald-de Waele fluid in laminar flow through a cylindrical tube. *Proc. Roy. Soc., Lond.* A 283, 576–582.
- SOUNDALEGEKAR, V. M. 1971 Effect of couple stresses in fluids on dispersion of a solute in a channel flow. *Phys. Fluids* 14, 19–20.
- TAYLOR, G. I. 1953 Dispersion soluble matter in solvent flowing slowly through a tube. *Proc. Roy. Soc., Lond.* A 219, 186–203.
- TAYLOR, G. I. 1954a The dispersion of matter in turbulent flow through a pipe. *Proc. Roy.* Soc., Lond. A 223, 446–468.
- TAYLOR, G. I. 1954b Conditions under which dispersion of a solute in a stream of solvent can be used to measure molecular diffusion. *Proc. Roy. Soc., Lond.* A 225, 473-477.